

19. OCEANIC BASALTS FROM THE TYRRHENIAN BASIN, DSDP LEG 42A, HOLE 373A

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ABSTRACT

During Leg 42A, drilling in Hole 373A in the Central Tyrrhenian Sea penetrated Miocene basaltic basement for 190 meters and recovered basalt from up to 450 meters below the sea floor. Despite mineralogical and chemical alteration due to hydrothermal processes in the Tyrrhenian marginal basin, petrological analogies to high-Al abyssal basalts can be traced.

Two types of interlayered basalt were recognized: (a) a lower Ti group (Core 5, Sections 1-2; Cores 7 and 10), and (b) a higher Ti group (Core 5, Section 3; Cores 11 and 12; and outer barrel). We distinguish two magma series: a more primitive magma leading to the plagioclase-bearing low-titanium basalts with affinities to high-Al abyssal tholeiites, and a more differentiated magma leading to the basalts with higher titanium content.

INTRODUCTION

During DSDP Leg 42A, basalt was recovered from Hole 373A in the abyssal plain of the Central Tyrrhenian Sea (Figure 1). Basaltic breccia was first encountered under a Quaternary and Pliocene sediment cover about 270 meters below the sea floor. Then it was drilled 190 meters through a basaltic sequence of breccia and lava flows and took 11 cores at depths as indicated on Figure 2.

Site 373A is on the lower flanks of one of the large Tyrrhenian seamounts (Figure 1) and the basalts belong to the acoustic basement (Site 373A report, this volume). Assuming that the age of the overlying sediments is its upper limit, the basaltic basement is probably lower Pliocene/upper Miocene (Barberi et al., 1977, Kreuzer et al., 1977).

GEODYNAMICS OF THE TYRRHENIAN SEA

The formation of small ocean basins in the western Mediterranean has caused considerable discussion (see Ritsema, 1970) since the discovery of their youthful age and the nature (oceanic to transitional) of the crust and mantle beneath became evident (Menard, 1967). Early geological work in the borderlands of the Tyrrhenian Sea had revealed the existence of a former landmass in the area presently occupied by the basin. Evidence of currents and clastic sedimentary components in the sedimentary sequence point to the existence of such a landmass until Miocene time. Following the earlier concept of the tectonic break-up of a Tyrrhenian continent, several other concepts involving complicated oceanization models were presented. These are reviewed by Ritsema (1970).

The plate tectonics concept invokes spreading processes with formation of new basaltic crust in

widening basins. In this context Boccaletti and Guazzone (1972), Dewey et al. (1973), and Barberi et al. (1973) have compared the Tyrrhenian Basin with formerly spreading but now inactive marginal basins which nevertheless still have high heat flow values (Karig, 1971).

The deep-focus earthquakes which occur in the southeastern Tyrrhenian Sea and the calcalkaline magmatism which occurs in the Aeolian Islands are characteristic of an active island arc (Barberi et al., 1973; Keller, 1974). The age of the exposed island arc volcanic rocks, however, is less than 1 m.y.B.P. (Barberi et al., 1974), whereas the Tyrrhenian Basin formed about 10 m.y. ago. The high heat flow values and a rather indistinct pattern of magnetic anomalies in the Tyrrhenian Basin suggest that it was a spreading marginal basin (Ryan et al., 1970). But the pattern of the Tyrrhenian spreading centers that can be inferred from the magnetic anomalies does not exhibit any simple regularity. The picture is certainly complicated by the rotation of individual crustal blocks such as Sardinia-Corsica, the Italian peninsula, and smaller microplates (Alvarez et al., 1974; Lowrie and Alvarez, 1974). Spreading centers may have developed between such rotating blocks and the existence of areas with continental crust within such a domain is possible. (Honnorez and Keller, 1968; Heezen et al., 1971).

The main purpose of drilling in the Tyrrhenian basement was to test whether the chemical characteristics of abyssal basalts, formed at oceanic spreading centers or their marginal basin analogs, are present. If so, we could conclude that upper mantle processes similar to those occurring under mid-ocean ridges have acted as well, in the Tyrrhenian Basin.

Dredge samples of basalt from the Tyrrhenian seamounts (mainly from Mount Marsili) were availa-

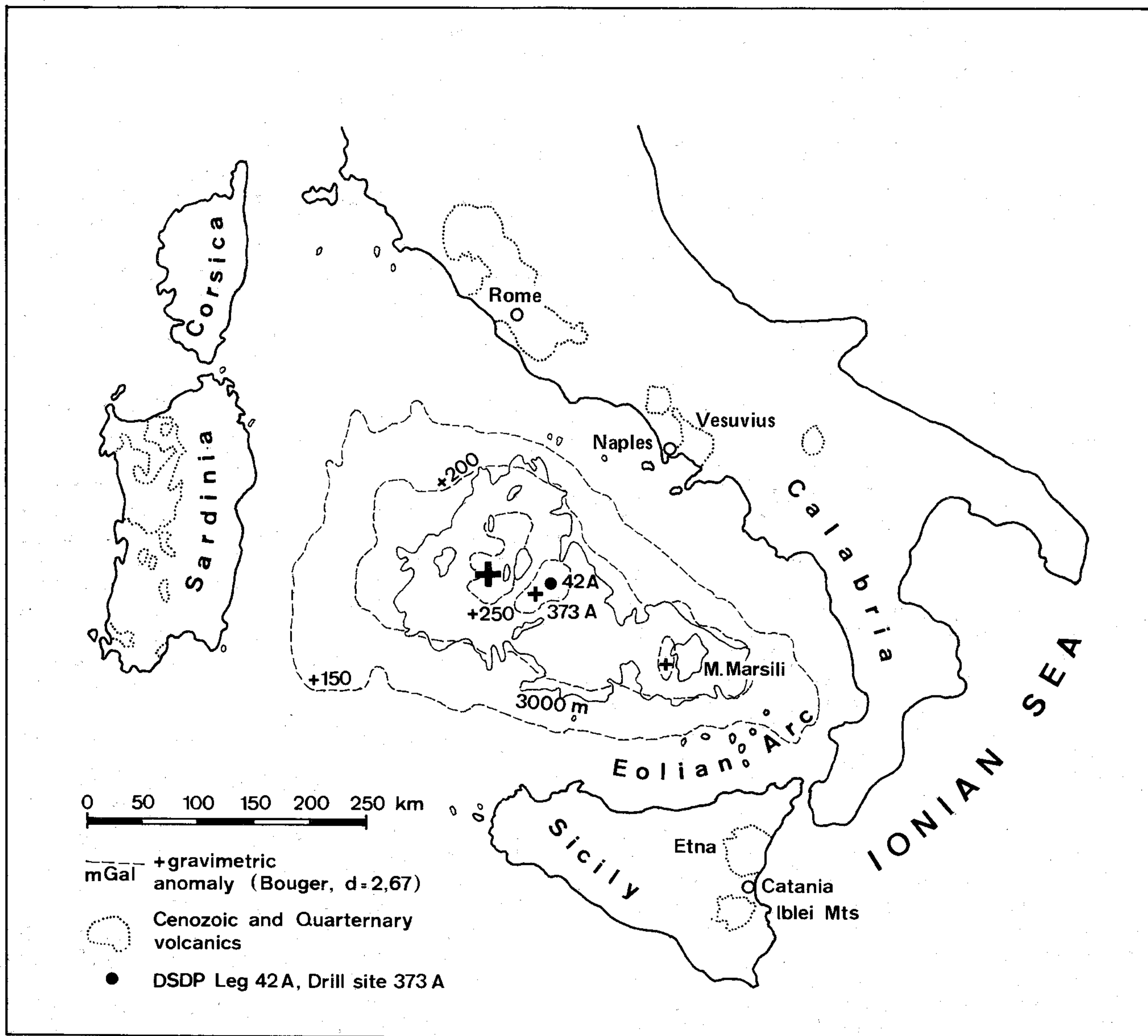


Figure 1. *The Tyrrhenian Basin.*

ble for study (Maccarone, 1970; Del Monte, 1972; Keller and Leiber, 1974) (Figure 1). These basalt samples are mostly alkaline. No data were published to support a more detailed characterization (for example, designation as "abyssal tholeiite") for there were few samples with tholeiitic affinity (e.g., No. 29M of Del Monte, 1972). Nevertheless, such designation was widely quoted to support the existence of spreading processes within the Tyrrhenian Sea. Moreover, seamount volcanism apparently occurred much later than basin formation and lasted until the upper Quaternary (Keller and Leiber, 1974).

PETROGRAPHIC DESCRIPTION OF ANALYZED SAMPLES

Most basalt samples are well crystallized and have intergranular textures of fresh plagioclase, clinopyroxene, and titanomagnetite in a matrix of smectite,

chloritic minerals, or both. This matrix constitutes up to 30% of the rock volumes. X-ray analyses, with use of ethylene glycol treatment, on three samples showed the typical shift of the 100 *montmorillonite* peak from 14 to 18 Å. A *chlorite* + *montmorillonite* mixture is abundant in two of the samples (7-2, 136-140 cm, and 7-3, 111-112.5 cm). Some smectite patches, clearly pseudomorphs after olivine and very rarely occurring olivine relics are preserved (e.g., 7-4, 135-140 cm, and 7-5, 10-14 cm). However, the extent of original modal olivine remains uncertain.

Two petrographic types of basalt are recognized; (1) plagioclase phyric and (2) aphyric to sparsely microphyric. This distinction corresponds to a chemical grouping explained below (Table 4).

Plagioclase phenocrysts are calcic with compositions ranging up to An_{84} . Groundmass plagioclase ranges mainly from An_{50-60} (Figure 3, Table 1). Along with

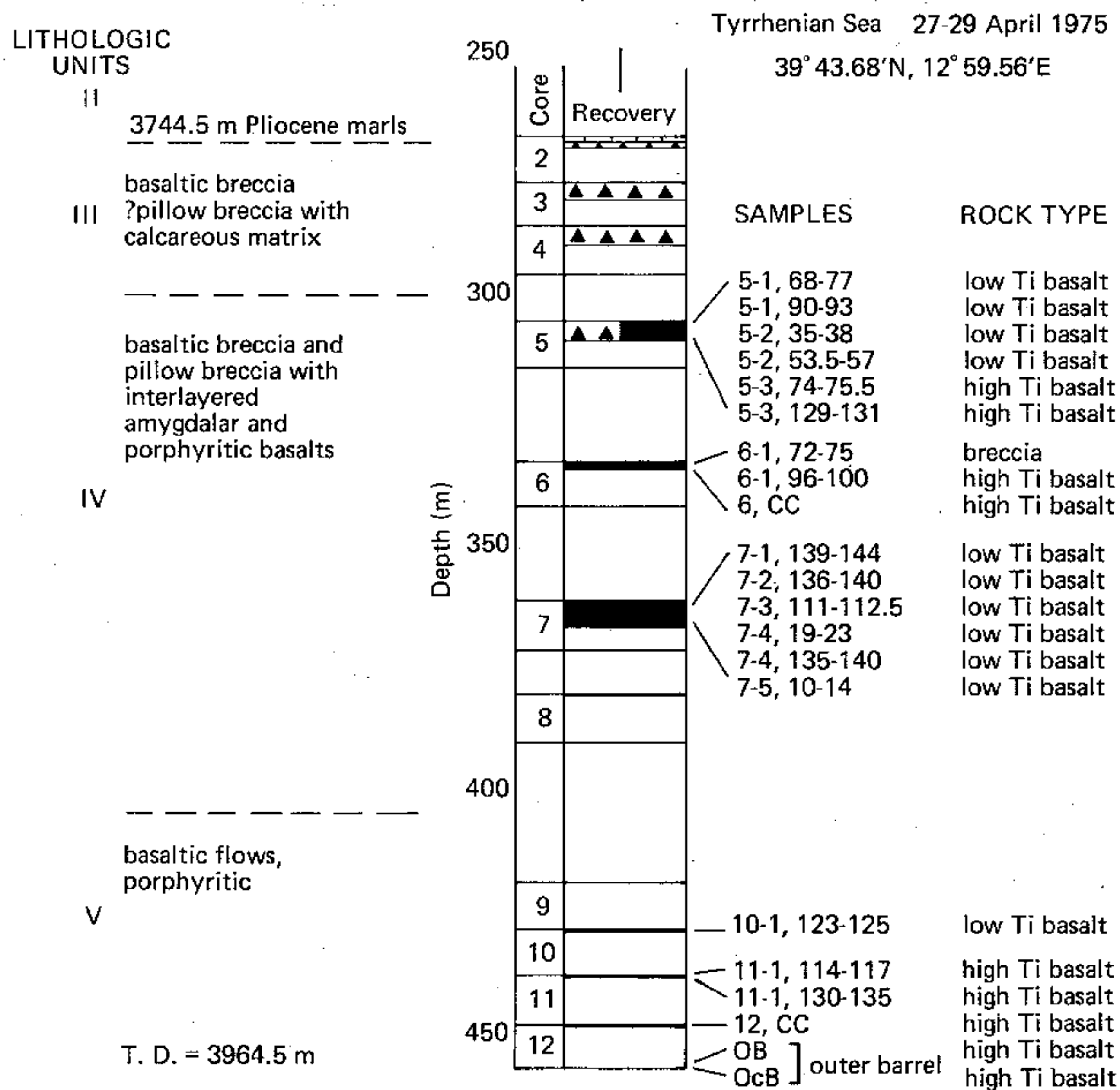


Figure 2. Graphic lithology of Site 373A. The drilling started at a depth of 96.5 meters sub-bottom within Quaternary Nannofossil marls (lithologic Unit I; Site 373A Report, this volume). At 270 meters, Glomar Challenger drilled into basaltic breccias (containing pillow fragments) associated with lava flows. The drilling was terminated at 457.5 meters in basaltic lava flows. Twenty basaltic samples from six cores out of Units IV and V were analyzed.

these groundmass labradorites, almost pure albite (with an increased orthoclase component) has been found by microprobe analysis. This suggests the beginning of a process of mineralogical readjustment and chemical exchange which is more pronounced in true ocean-floor metamorphic spilites (Cann, 1969).

Chlorite and an unidentified green amphibole-like mineral (riebeckite?) developed as small rims surrounding the groundmass clinopyroxene and as tiny, but discrete, crystals within the chlorite/smectite patches. X-ray powder data from Sample 7-4, 10-4 cm, confirmed the existence of chlorite. Thus, the first indication of a change toward a greenschist facies

mineralogy as described for oceanic basalts by Miyashiro et al. (1971) is apparent.

In general, however, *groundmass clinopyroxenes* are fresh. Their composition (Table 2, Figure 4) is salitic and similar to the clinopyroxenes reported by Ridley et al. (1974). Iron enrichment is present but not pronounced. No Ca-poor pyroxene has been detected.

Calcite occurs as veinlets, vesicle fillings, and patches within the groundmass and plagioclase phenocrysts. All samples analyzed contain only minor amounts of calcite. Zeolites replace plagioclase and fill vesicles in the most altered samples, but these are not suitable for chemical analysis (5-1, 68-71 cm; 5-2, 35-38 cm; and 6-1, 133-136 cm).

GEOCHEMICAL METHODS AND DISCUSSION OF RESULTS

Twenty samples were analyzed by X-ray fluorescence, atomic absorption, ionic neutron activation, and wet chemical methods to determine their major and trace element contents. All major element determinations were made at least three times by each of the different laboratories involved (Freiburg i.Br., Karlsruhe, Zürich, and Canberra). The results given in Table 3 are in good agreement. Determinations were carried out by:

- XRF for SiO_2 , TiO_2 , Al_2O_3 , Fe_2O_3 (total), MnO , MgO , CaO , and K_2O ;
- AAS for Na_2O , K_2O , CaO , MgO , Fe_2O_3 (total), and Al_2O_3 ;
- INAA for Na_2O , K_2O , CaO , MnO , and Fe_2O_3 (total);
- Wet chemical methods for SiO_2 , TiO_2 , FeO , P_2O_5 , H_2O^+ , and CO_2 ;
- Electron microprobe analysis (by J. Keller) on plagioclase and pyroxene phenocrysts.

All samples were analyzed for the following elements by INAA, but these elements were below the detection limits indicated:

Sb (<1 ppm), Ta (<1 ppm), U (<1.5 ppm), Th (<1 ppm), and Cs (<2 ppm). The trace element data are included in Table 3. The methods applied were: (a) XRF for Rb, Sr, Zr, Y, Pb, and U; (b) AAS for Li, Sr, and Cu; (c) INAA for Sc, Cr, Co, Hf, La, Ce, Sm, Eu, Tb, Dy, Yb, and Lu

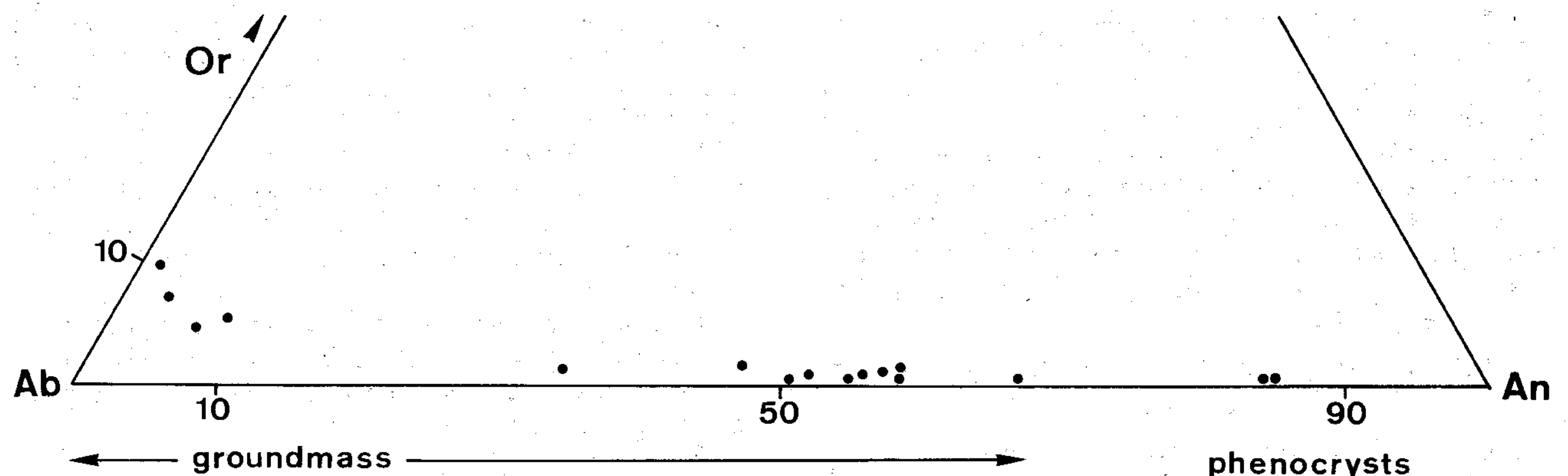


Figure 3. Plagioclase plot for Hole 373A basalts.

TABLE 1
Chemical Composition and Structural Formula of Phenocrystic and Groundmass Plagioclases from 373A Basalts (microprobe analyses)

	5-1, 90-93 cm		5-2, 53.5-57 cm		6, CC	7-1, 139-144 cm		11-1, 130-135 cm		12, CC							
	Grdm.	Pheno.	Grdm.	Grdm.	Grdm.	Grdm.	Grdm.	Pheno.	Grdm.	Grdm.	Grdm.	Grdm.	Grdm.	Grdm.	Grdm.	Grdm.	
SiO ₂	56.66	45.75	50.77	54.88	58.61	54.56	65.16	52.84	45.97	53.11	53.42	53.65	53.10	53.32	66.22	66.75	67.76
Al ₂ O ₃	27.02	34.72	30.97	28.56	25.06	28.28	20.55	29.46	34.29	29.32	29.13	29.02	29.83	29.19	20.61	20.08	18.90
FeO total	0.86	0.24	0.6	0.99	1.06	0.90	1.27	0.68	0.33	0.63	1.67	0.66	1.43	0.64	0.42	0.57	0.75
CaO	9.05	17.46	13.71	9.44	7.15	10.57	1.71	12.15	17.38	11.82	11.08	11.35	10.19	11.54	1.32	0.65	0.31
Na ₂ O	6.41	1.70	3.85	5.82	7.62	5.53	10.19	4.77	1.75	4.89	4.44	5.17	5.28	5.13	10.55	10.68	10.53
K ₂ O		0.13	0.09	0.31	0.32	0.16	1.00	0.09	0.15	0.23	0.26	0.16	0.17	0.19	0.88	1.26	1.76
32 O																	
Si	10.203	8.437	9.269	9.920	10.541	9.883	11.568	9.560	8.483	9.649	9.719	9.731	9.639	9.682	11.679	11.778	11.965
Al	5.733	7.546	6.663	6.085	5.312	6.038	4.300	6.315	7.458	6.279	6.239	6.203	6.383	6.247	4.284	4.177	3.933
Fe	0.129	0.037	0.092	0.150	0.159	0.137	0.189	0.104	0.050	0.096	0.253	0.099	0.217	0.097	0.062	0.085	0.111
Ca	1.746	3.451	2.682	1.827	1.378	2.051	0.325	2.368	3.436	2.301	2.158	2.206	1.983	2.244	0.250	0.124	0.059
Na	2.238	0.609	1.364	2.041	2.657	1.942	3.509	1.682	0.628	1.721	1.566	1.817	1.857	1.807	3.608	3.655	3.604
K		0.031	0.022	0.071	0.074	0.037	0.226	0.020	0.035	0.053	0.060	0.037	0.040	0.043	0.198	0.284	0.396
	20.049	20.111	20.092	20.094	20.121	20.088	20.117	20.049	20.090	20.099	19.995	20.093	20.119	20.120	20.081	20.103	20.068
Na:	56.2	14.9	33.5	51.8	64.4	48.2	86.4	41.3	15.3	42.2	41.4	44.8	47.9	44.1	89.0	90	88.8
K: Mol%	0	0.8	0.5	1.8	1.8	0.9	5.6	0.5	0.8	1.3	1.6	0.9	1.0	1.0	4.9	7	9.8
Ca:	43.8	84.4	65.9	46.4	33.5	50.9	8.0	58.2	83.8	56.5	57.0	54.3	51.1	54.8	6.2	3	1.4

TABLE 2
Chemical Composition and Structural Formula of Groundmass Clinopyroxenes from 373A Basalts (microprobe analyses)

	5-1, 90-93 cm		5-2, 53.5-57 cm		6, CC		7-1, 139-144 cm		11-1, 130-135 cm		12, CC					
SiO ₂	49.72	48.61	48.21	50.30	50.45	50.05	50.20	51.99	50.33	51.06	48.89	52.62	52.35	49.04		
TiO ₂	1.03	1.67	1.62	0.91	1.04	1.19	0.90	0.59	0.84	0.95	1.81	0.58	0.73	1.59		
Al ₂ O ₃	5.75	4.43	4.32	5.00	3.87	4.29	4.50	3.29	4.68	3.62	4.32	1.91	2.47	4.83		
FeO total	6.47	10.80	13.35	5.71	7.61	7.44	6.52	5.31	5.80	7.12	9.93	7.69	7.21	7.90		
MnO		0.15	0.26				0.13				0.12			0.12		
MgO	14.98	13.58	11.48	15.06	14.75	14.81	15.14	15.87	15.13	15.02	13.23	16.42	16.20	14.44		
CaO	20.92	20.65	20.09	22.11	21.58	21.37	21.73	22.22	21.96	21.72	20.83	20.32	20.60	20.90		
Na ₂ O	0.41	0.59	0.55	0.48	0.46	0.36	0.27		0.32	0.31	0.62	0.28	0.22	0.55		
Cr ₂ O ₃	0.65	0.23	0.13	0.43	0.24	0.36	0.75	0.73	0.94	0.20	0.24	0.18	0.22	0.64		
Si	= 2.0 {	1.836	1.831	1.840	1.855	1.877	1.859	1.856	1.908	1.858	1.890	1.840	1.941	1.928	1.829	}
Al		0.164	0.169	0.160	0.145	0.123	0.141	0.144	0.092	0.142	0.110	0.160	0.059	0.072	0.171	
Al	~ 2.0 {	0.086	0.031	0.034	0.072	0.046	0.047	0.052	0.050	0.062	0.048	0.032	0.024	0.035	0.041	}
Fe		0.200	0.318	0.426	0.176	0.237	0.231	0.202	0.163	0.179	0.221	0.312	0.237	0.222	0.246	
Mn		0.005	0.008				0.004				0.004			0.004		
Mg		0.825	0.762	0.653	0.828	0.817	0.820	0.835	0.868	0.833	0.824	0.742	0.903	0.889	0.803	~ 2.0
Ca		0.827	0.833	0.822	0.874	0.859	0.850	0.861	0.874	0.868	0.861	0.840	0.803	0.813	0.835	
Na		0.030*	0.043	0.041	0.034	0.033	0.026	0.019		0.023	0.023	0.045	0.020	0.015	0.040	
Cr		0.019	0.007	0.004	0.013	0.007	0.011	0.022	0.021	0.027	0.005	0.007	0.005	0.007	0.019	
Ti		0.029	0.047	0.047	0.025	0.029	0.033	0.025	0.016	0.023	0.026	0.051	0.016	0.020	0.045	
		*+0.003 K														
Mg:	44.5	39.8	34.3	44.1	42.7	43.1	44.0	45.6	44.3	43.4	39.2	46.5	46.2	42.6		
Ca: Mol%	44.7	43.5	43.2	46.5	44.9	44.7	45.4	45.9	46.2	45.1	44.3	41.3	42.2	44.3		
Fe:	10.8	16.6	22.4	9.4	12.4	12.1	10.6	8.6	9.5	11.5	16.5	12.2	11.5	13.1		

To check accuracy, standard reference samples were also analyzed. The values obtained for USGS-BCR-1 in the Karlsruhe laboratory are given by Puchelt et al. (1976) in the Initial Reports of the Deep Sea Drilling Project, Volume 37.

In addition, the abundance of a large number of trace elements was determined by XRF, AAS, and INAA. A few samples were tested by Spark Source Mass Spectroscopy for Rb, Sr, Ba, Pb, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Yb, Lu, Y, Nb, Zr, Th, and U.

A close scrutiny of Table 3 reveals that the basalts differ markedly in their TiO₂ contents and may be divided into two groups, the averages of which are significantly different. Accordingly, we have discerned lower Ti and higher Ti basalts. Once the grouping is made, further distinctions become obvious for other

elements. To obtain a suitable basis for comparison, the actual analytical data have been recalculated on a H₂O and CO₂-free basis. The averages for the two groups are listed together with the standard deviation in Table 4. Along with the quite pronounced differences in TiO₂, a similar clustering of values occurs for Al₂O₃, Na₂O, and P₂O₅. Their respective concentrations versus TiO₂ are shown in Figure 5. The higher Ti group is enriched in SiO₂, Fe₂O₃, Na₂O, P₂O₅, and rare earth element (REE) concentrations, whereas the lower Ti group is enriched in Al₂O₃, MgO, and CaO.

All basalts from Site 373A show secondary hydration and oxidation (Figure 6). Samples 5-1, 68-75cm and 5-2, 35-38 cm, are high in CaCO₃.

On an alkali-silica diagram, the Hole 373A basalts fall close to the alkaline field boundary or even in the high-alumina basalt field (Figure 7). Most samples are

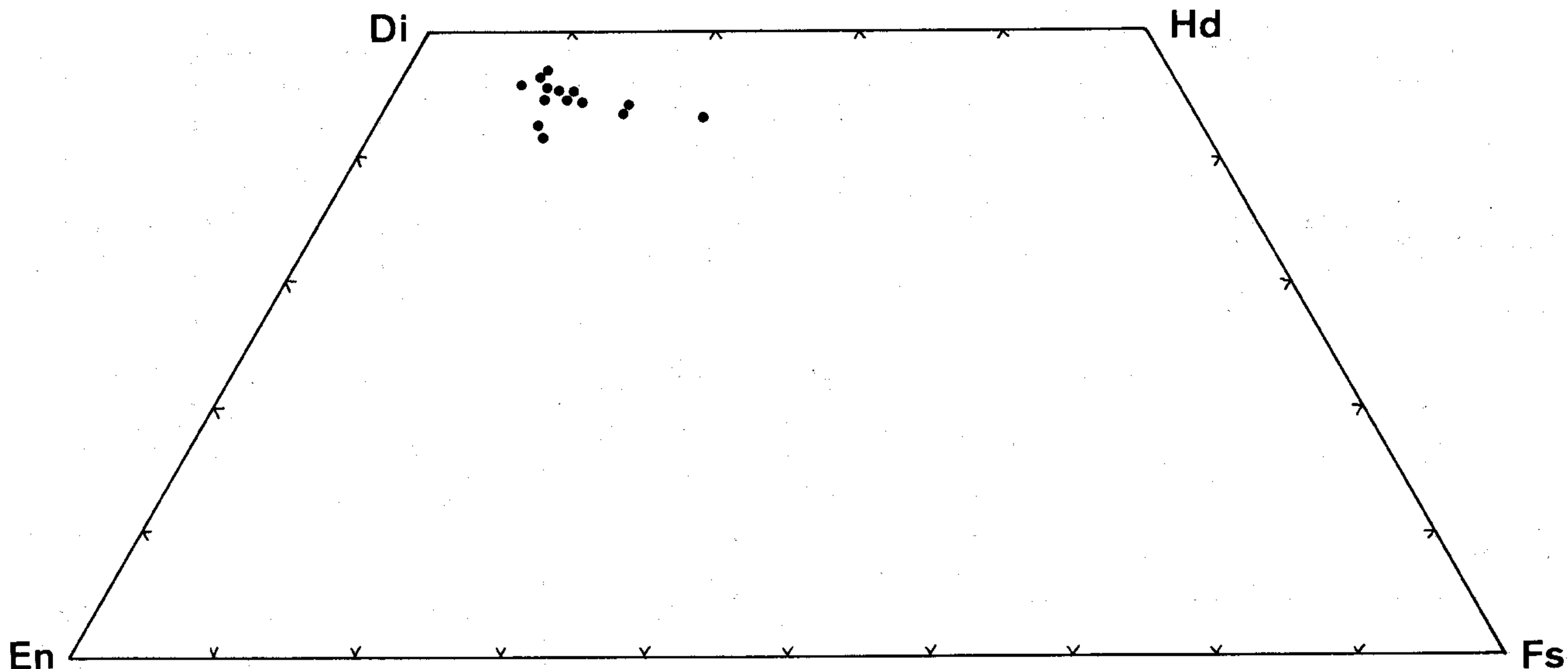


Figure 4. Clinopyroxene plot for Hole 373A basalts.

nepheline-normative (Table 5 and Figure 8) using a corrected iron oxidation of $\text{Fe}_2\text{O}_3/\text{FeO} = 0.15$. This may not reflect the original chemistry. Na_2O could have been introduced (spilitization). In some of these samples the content of CaO is 7-8%. This is well below oceanic tholeiite concentrations whose range is exemplified, for example, in the $\text{CaO}/\text{Al}_2\text{O}_3$ plot of Kay et al. (1970). Miyashiro et al. (1971) reported considerable loss of CaO with increase of Na_2O in metabasalts of the Mid-Atlantic Ridge.

On the other hand, the higher Ti basalts show a consistent enrichment of SiO_2 , Na_2O , Fe_2O_3 , P_2O_5 , and REE.

The data suggest that the two groups represent two initially different basalt types (Figures 7, 9, 10, 11, 12, 13, and 14). The higher Ti basalts have higher normative contents of the Fe-Ti-oxides and apatite, and lower contents of olivine (Table 5 and Figure 8). Each type has a different degree of fractionation (Figure 10). The MgO/FeO ratio expressed as Mg-numbers (molecular $100 \text{ Mg}/\text{Mg} + \text{Fe}^{+2}$ (Green and Ringwood, 1967). The Fe^{+2} , standardized as $\text{Fe}_2\text{O}_3/\text{FeO} = 0.15$, shows a broad range of fractionation with Mg-numbers from 74 to 46 (Table 5). The higher Ti basalts (Mg-numbers: 46-66 and differentiation index: 33-41) could represent a more fractionated part of the magma than the lower Ti basalts. The highest Mg-numbers (65-74 for the lower Ti basalts) are consistent with a direct partial melting from upper mantle peridotite (Green, 1970). Magma segregation following about 15% partial melting at about 9 kb at a depth range of 15-35 km would yield a high-alumina olivine tholeiite of appropriate composition. The low TiO_2 , very high Al_2O_3 , and high normative olivine content (up to 20%) may require a refinement of the model with a greater degree of partial melting in a Al-rich source. The lower Ti basalts could be plagioclase-tholeiites of Shido et al. (1971). However, this terminology is not adopted, because the

alteration of olivine obscures the crystallization sequence.

TRACE ELEMENTS

Trace element concentrations are given in Table 3. Chondrite normalized REE patterns are shown in Figure 14.

Concentrations of large-ion-lithophile (LIL) elements (such as Rb, Sr, Ba, Y, Zr, Hf, U) are low, indicating that these basalts are similar to abyssal tholeiites. Correlation plots of alteration-resistant element pairs, such as Zr-Sr, Zr-Nb, or P_2O_5 - TiO_2 (not illustrated), clearly fall in the field of oceanic ridge basalts (Bass et al., 1973; Ridley et al., 1974). Zr-Ti and Y-Ti correlations also fall on the positive correlation line outlined for ocean floor basaltic rocks by Cann (1970). This emphasizes that Zr and Y have slightly lower than average concentrations in the Tyrrhenian lower Ti basalts, and that they are markedly higher in the higher Ti basalts. La and Hf also are different in the two basalt types previously distinguished (Figure 13).

Rb, Sr, Ba, Pb, U, and Th have concentrations slightly but consistently above those found in ocean floor basalts (Engel et al., 1965; Kay et al., 1970; Hart, 1971; Church and Tatsumoto, 1975).

Relative to chondrites, the shape of the rare earth patterns and the Eu anomalies (Figure 14) are interesting. All samples have a similar general REE pattern. They are light REE enriched with $(\text{La}/\text{Sm})_{\text{e.f.}}$ of about 1.5. The enrichment of the heavy REE shows differences which coincide with the grouping into the lower Ti and higher Ti groups of basalts (Table 4 and Figures 7-13). This high $(\text{La}/\text{Sm})_{\text{e.f.}}$ is a distinctive feature of our REE patterns and is at variance with abyssal tholeiites which typically have $(\text{La}/\text{Sm})_{\text{e.f.}} < 1$ (Schilling, 1971).

TABLE 3
Chemical Composition of the Leg 42A Hole 373A Basalts, (n.d. = not determined)

	(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)	(10)	(11)	(12)	(13)	(14)	(15)	(16)	(17)	(18)	(19)	(20)
DSDP	5-1	5-1	5-2	5-2	5-3	5-3	6-1	6cc	7-1	7-2	7-3	7-4	7-4	7-5	10-1	11-1	11-1	12cc	OB 2	Ocb
Leg 42A-373A	68-71	90-93	35-38	53.5-57	74-75.5	129-131	96-100		139-144	136-140	111-112.5	19-23	135-140	10-14	123-125	114-117	130-135		Outer	barrel
SiO ₂	40.9	49.2	41.5	47.1	48.1	48.4	51.1	49.2	48.0	47.8	47.7	44.5	43.2	44.2	49.2	51.0	50.5	50.0	50.1	50.6
TiO ₂	1.12	1.08	1.07	1.03	1.55	1.71	1.75	1.41	0.92	0.84	1.02	0.84	0.91	1.09	1.11	1.68	1.57	1.56	1.72	1.86
Al ₂ O ₃	17.8	18.7	17.9	18.9	17.4	16.1	17.1	16.1	18.3	20.2	18.2	20.3	18.0	18.2	18.1	16.5	16.9	16.4	16.8	16.3
Fe ₂ O ₃	5.12	4.48	5.40	5.29	6.89	6.61	7.16	5.63	5.19	4.82	5.28	5.06	5.54	5.9	4.24	5.65	4.71	4.81	4.98	5.20
FeO	2.78	2.80	2.18	2.30	2.95	3.38	2.98	3.84	2.65	4.95	2.81	2.50	2.51	3.30	2.47	4.29	4.50	4.03	4.23	4.33
MnO	0.23	0.11	0.17	0.11	0.19	0.19	0.17	0.16	0.10	0.13	0.16	0.15	0.17	0.27	0.12	0.17	0.16	0.16	0.18	0.16
MgO	7.03	7.03	7.9	7.60	4.83	5.72	4.02	7.54	7.66	7.56	8.35	8.65	10.3	9.58	8.76	7.05	7.95	8.13	7.21	7.01
CaO	14.1	10.5	12.9	11.4	10.4	10.2	9.5	10.0	10.3	11.17	11.8	10.8	11.1	10.3	9.24	7.57	7.3	7.0	7.70	7.23
Na ₂ O	3.02	2.93	2.43	2.97	3.84	4.22	4.65	3.61	3.16	2.61	2.73	2.83	2.53	2.27	3.24	4.13	4.28	4.40	4.18	4.44
K ₂ O	0.32	0.30	0.28	0.32	0.29	0.43	0.38	0.32	0.33	0.27	0.21	0.24	0.23	0.25	0.27	0.27	0.28	0.32	0.32	0.37
P ₂ O ₅	0.20	0.17	0.16	0.12	0.26	0.23	0.25	0.19	0.12	0.10	0.13	0.12	0.11	0.14	0.14	0.24	0.22	0.22	0.26	0.26
H ₂ O	3.71	2.49	4.80	2.40	1.88	2.38	0.92	1.97	3.39	2.51	2.41	3.22	4.44	4.58	3.00	2.27	2.14	2.31	2.12	2.56
CO ₂	4.42	<0.1	4.12	0.1	0.1	<0.1	<0.1	<0.1	<0.1	0.1	<0.1	<0.1	1.5	0.8	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
Total	100.75	99.8	100.8	99.54	98.63	99.57	99.98	100.02	100.12	99.8	100.89	99.21	100.54	100.89	99.89	100.22	100.51	99.34	99.80	100.32
Li	10	29	33	32	18	27	9	29	29	27	29	34	50	46	34	11	17	11	8	15
Rb	<1	<1	<1	<1	<1	<1	<1	<2.5	<2.5	<1	<1	3	2	<1	<1	<1	<1	<1	<1	<1
Sr	200	250	225	215	230	220	160	205	232	215	225	237	215	235	210	180	205	155	197	230
Ba	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	58.5	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	58	n.d.	n.d.	n.d.
Pb	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	2.2	n.d.	n.d.	1.5	1.5	n.d.	n.d.	n.d.	2.4	n.d.	n.d.	n.d.
Cu	68	76	86	76	98	72	26	84	61	66	69	67	84	150	105	84	70	68	76	72
Co	29	32	33	29	37	31	32	35	30	39	33	34	31	39	35	32	32	30	31	31
Cr	201	199	206	207	190	199	169	207	227	285	255	253	210	210	221	138	133	142	115	100
Sc	34	32	31	29	39	35	36	43	29	32	33	30	33	34	35	32	34	33	32	29
Y	n.d.	28	28	n.d.	n.d.	37	n.d.	28	23	n.d.	22	18	21	26	28	40	39	n.d.	43	36
La	5.8	5.6	5.2	6.2	7.3	9.3	9.1	6.1	6.3	3.7	4.7	6.1	5.3	4.5	5.2	9.3	9.1	9.2	8.7	10.0
Ce	14.0	10.9	11.5	11.9	16.0	20.1	22.9	15.1	15.0	8.9	11.8	13.0	12.0	11.1	12.2	20.0	20.2	20.3	18.9	20.5
Sm	2.6	1.9	2.0	2.3	3.2	3.5	3.6	3.2	1.9	1.8	2.1	2.3	3.4	2.3	2.2	2.9	3.4	3.7	3.8	4.2
Eu	0.86	1.04	0.82	1.10	1.44	1.43	1.37	1.25	0.90	0.84	0.89	0.92	0.71	0.85	0.98	1.43	1.24	1.34	1.30	1.57
Tb	1.1	<1	<1	<1	1.5	1.5	1.2	1.0	0.63	<1	<1	<1	<1	1.1	<1	1.6	<1	<1	1.2	<1
Dy	3.2	3.0	3.4	3.2	5.2	4.5	5.1	4.7	<1	2.6	3.2	2.8	3.2	3.2	3.5	4.8	5.6	5.2	5.9	5.3
Yb	2.2	1	1.6	2.4	4.4	3.5	2.8	2.9	2.2	1.5	1.8	<1	1.1	2.2	2.3	3.0	3.4	3.5	2.8	2.5
Lu	0.27	0.26	0.28	0.27	0.40	0.43	0.37	0.38	0.29	0.26	0.31	0.28	0.25	0.30	0.30	0.46	0.49	0.42	0.46	0.50
Zr	n.d.	81	82	n.d.	n.d.	142	n.d.	107	77	n.d.	69	61	54	79	90	156	140	n.d.	165	153
Hf	2.2	<1	2.0	0.9	2.7	2.7	3.0	2.2	1.6	1.9	2.0	1.5	1.2	2.0	1.5	3.3	3.2	3.3	2.4	3.1
Nb	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	3.1	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	6.5	n.d.	n.d.	n.d.
Ta	<1	<1	<1	<1	0.85	<1	<1	0.77	<1	<1	<1	<1	<1	0.95	<1	<1	1.08	<1	<1	<1
Th	<3	<3	<3	<3	<3	<3	<3	<3	0.42	<3	<3	<3	<3	<3	<3	<3	0.63	<3	<3	<3
U	<2	<2	<2	<2	<2	<2	<2	<2	0.24	<2	<2	0.11	0.21	<2	<2	<2	0.8	<2	<2	<2

TABLE 4
Chemical and Petrographical Characteristics of
the Two Basalt Types of Hole 373A on a
(dry) Volatile Free Basis

	Lower Ti Group (11 samples) $\bar{X} \pm 1 S$	Higher Ti Group (9 samples) $\bar{X} \pm 1 S$
SiO ₂ (Wt %)	47.8 ± 2.2	51.0 ± 1.1
TiO ₂	1.05 ± 0.12	1.68 ± 0.14
Al ₂ O ₃	19.4 ± 0.8	17.0 ± 0.4
Fe ₂ O ₃ Total	8.65 ± 0.97	10.2 ± 0.42
MnO	0.16 ± 0.05	0.17 ± 0.01
MgO	8.60 ± 1.13	6.75 ± 1.47
CaO	11.8 ± 1.6	8.74 ± 1.44
Na ₂ O	2.92 ± 0.30	4.29 ± 0.31
K ₂ O	0.29 ± 0.03	0.34 ± 0.04
P ₂ O ₅	0.14 ± 0.03	0.24 ± 0.03
Σ REE (ppm)	40.1 ± 4.2	64.4 ± 5.4
Cr (ppm)	255 ± 36	155 ± 27
Mg number	65 - 74	46 - 66
Differentiation Index	20 - 29	33 - 41
	plagioclase phyric	mainly a phyric

The positive Eu anomaly is obviously linked to samples with high Al₂O₃ contents, indicating that plagioclase is the Eu-bearing phase.

PETROGENETIC DISCUSSION

Drilling on Leg 42A encountered, for the first time, abyssal basalts in the young Mediterranean deep-sea basins. Two significantly different and interlayered types of basalts were recovered from Hole 373A; *lower Ti basalts* with affinities to high-alumina abyssal tholeiites, and *higher Ti basalts*.

Alteration of these basalts is widespread and involves chemical migrations. It is most evident in the CaCO₃ bearing samples and in apparently Na-enriched and Ca-depleted samples (spilitization). The ocean floor metamorphic processes responsible are probably related to circulation of hydrothermal solutions in areas of higher heat flow; they merit further study. At present, we concentrate primarily upon the recognition of original magma types.

According to Green and Ringwood (1967, 1969), 01-normative, high-alumina tholeiite (the Site 373A low-Ti basalts) is generated at a relatively shallow mantle depth (about 9 kb or a model depth of 15-35 km) and by high degree of partial melting (25%-30%). The origin of basalt within these limitations is thought to be related to upwelling or intrusion of partly molten mantle material from the low-velocity zone into shallower levels of a narrow axial zone at spreading centers. High heat flow is related to the rising, hot mantle material. Basalts of the lower Ti group with highest Mg numbers can be directly related to this model and do not require large-scale fractionation processes to account for their composition. Plagioclase accumulation can explain the extremely high Al₂O₃ enrichment and a possibly existing Eu anomaly related to high Sr contents.

Variations in LIL and other incompatible elements between abyssal basalts from different regions may

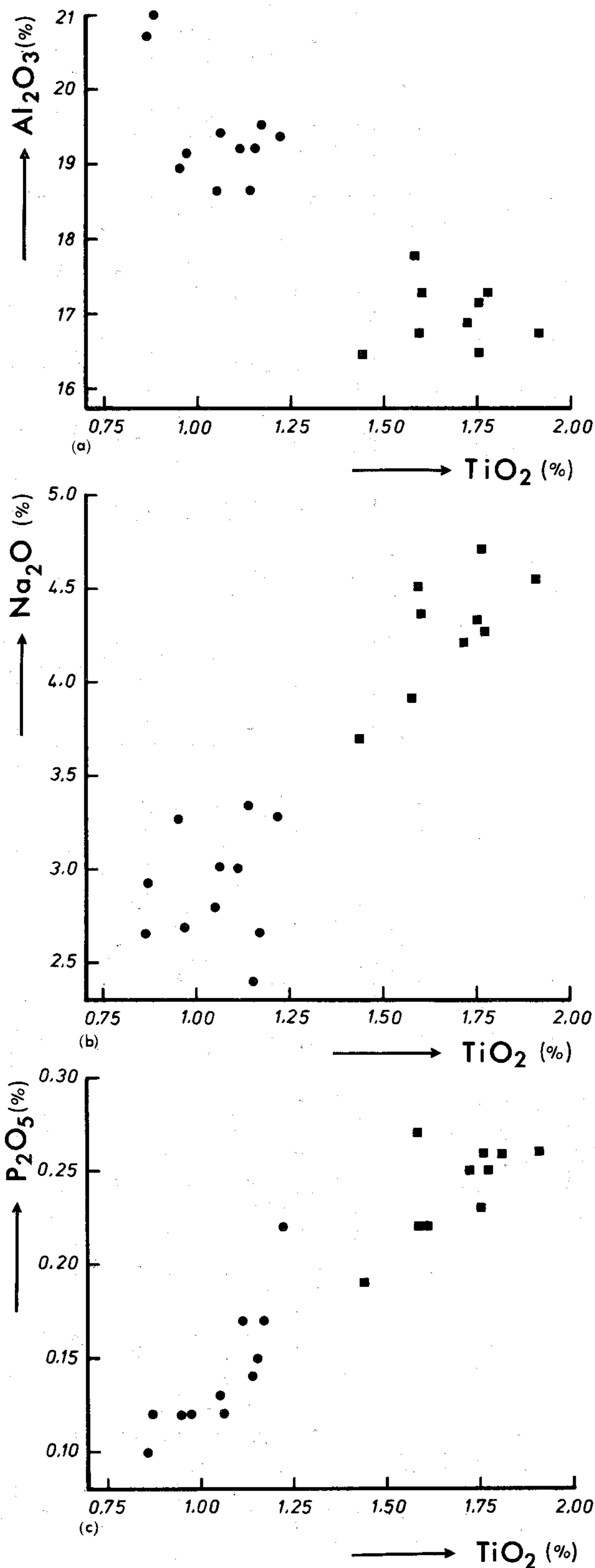


Figure 5. Variation diagrams for Hole 373A basalts. (a) Al₂O₃/TiO₂, (b) Na₂O/TiO₂; (c) P₂O₅/TiO₂. Solid circles = low-Ti basalts; solid squares = high-Ti basalts. For location see Figure 2.

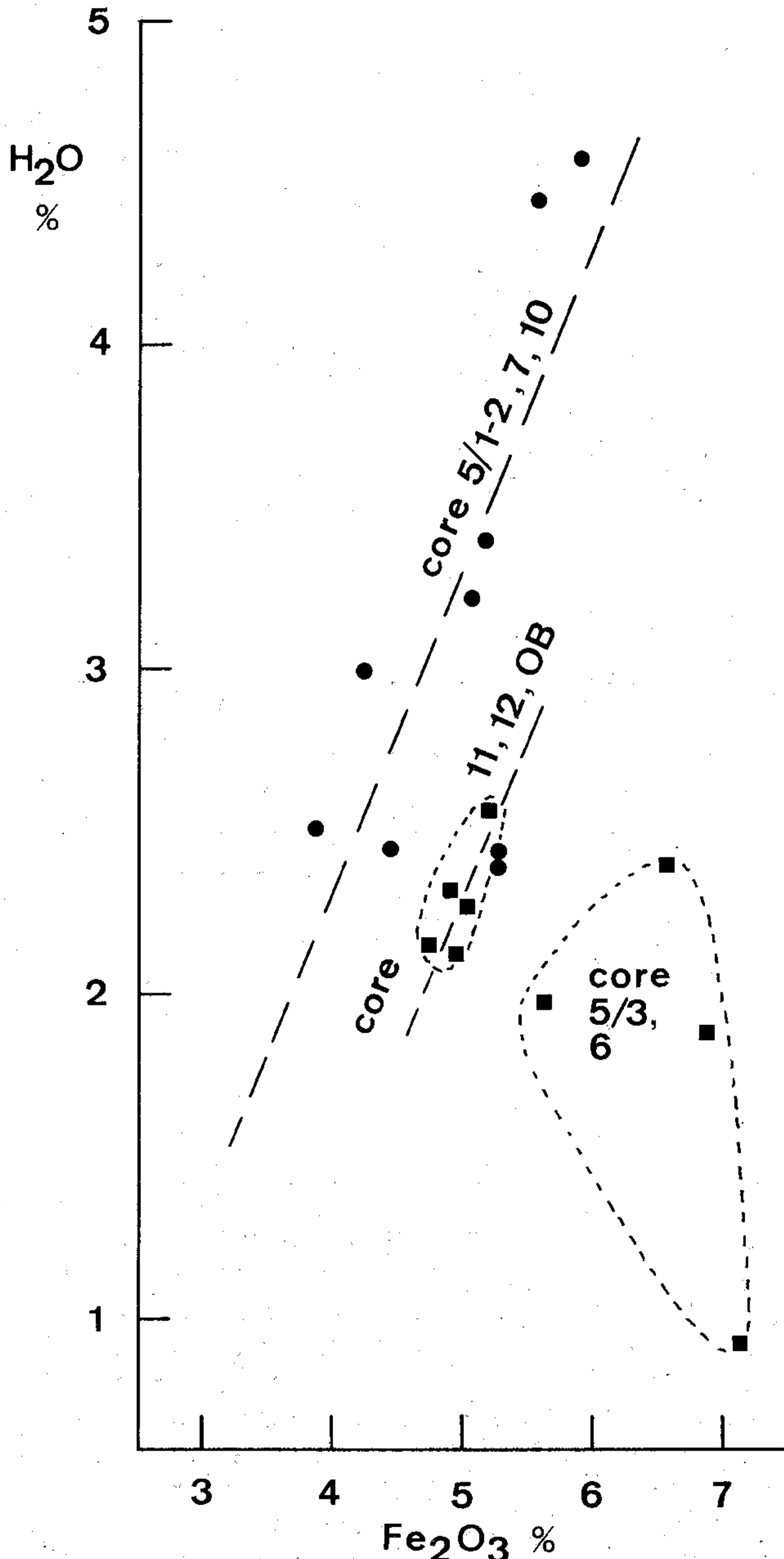


Figure 6. Increase of Fe_2O_3 and H_2O within the 373A basalts indicating the degree of weathering and metamorphism. Solid circles = Ti basalts, solid squares = high Ti-basalts.

reflect different concentrations in the source rocks. The source of the most unfractionated Tyrrhenian basalts yields lower Ti, Y, Zr and is enriched in light REE, Ba, Rb, Sr, Th, U, Pb, compared with oceanic basalts. Kay et al. (1970) showed that extremely low LIL element concentrations along the Mid-Atlantic Ridge resulted from depletion during previous partial melting episodes. The depletion under the Tyrrhenian Basin is less pronounced.

A postulated vertical zoning of the upper mantle, reflected in the enrichment of LIL elements in the upper levels of the low velocity zone (Green, 1970), can also lead to increased LIL element concentrations

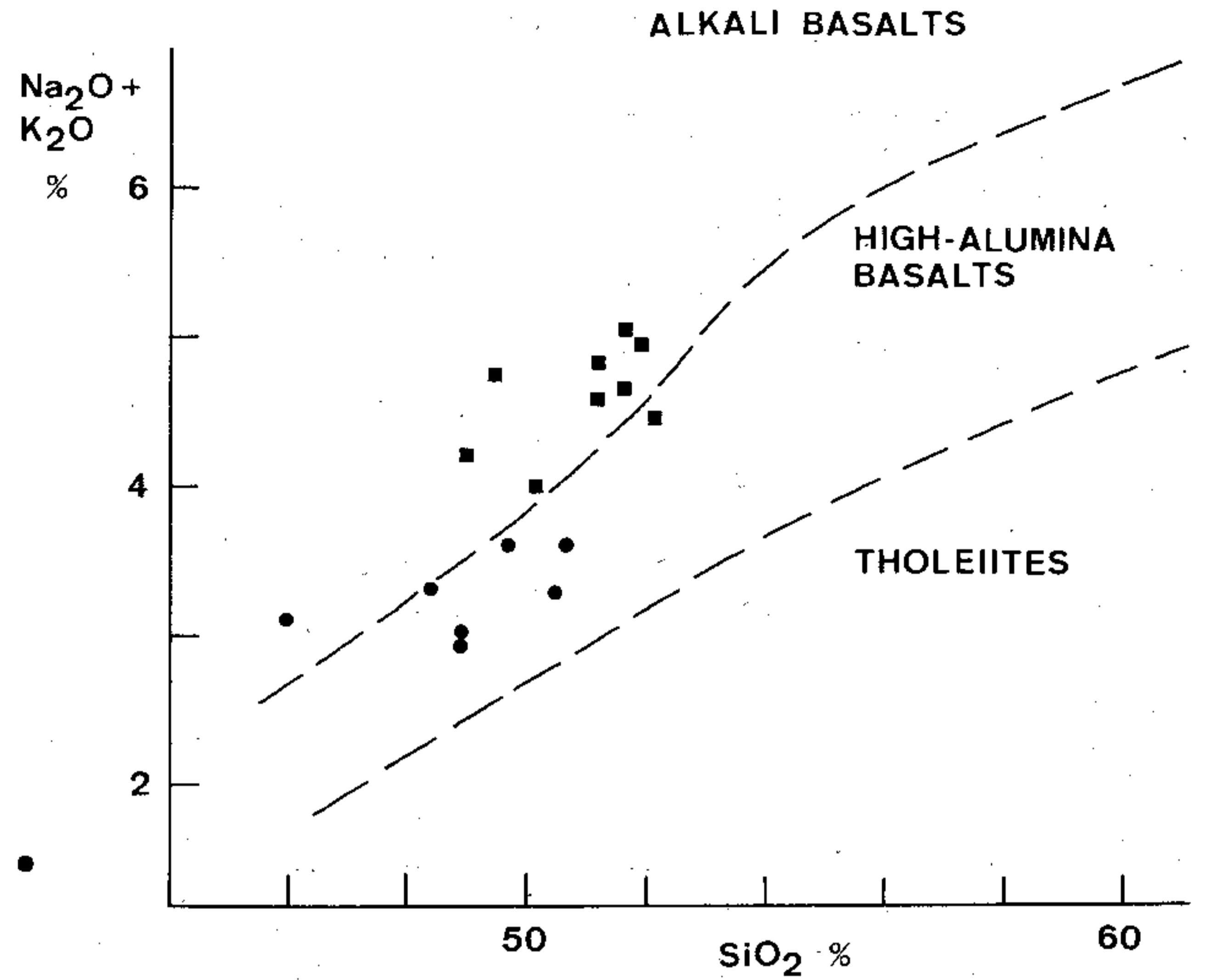


Figure 7. Alkali/ SiO_2 plot on water-free basis (solid circles = low-Ti basalts, solid squares = high-Ti basalts).

in partial melts. If the upper mantle below the Tyrrhenian Sea has a structure similar to that beneath the Balearic Basin (Berry and Knopoff, 1967), then an ultra-low velocity channel upwarp to only 50 km below the central basin could well provide higher LIL elements even in tholeiites representing high degrees of partial melting.

The Hole 373A higher Ti basalts could have originated as Qz-normative basalts (containing lower Na_2O contents, averaging approximately 2.8 wt.%). Such original rocks apparently were more fractionated than the lower Ti basalts. No simple fractionation line relates the two groups. To account for the higher Ti basalts, we envision a complex origin involving a lesser degree of partial melting (to explain higher Ti, Hf, Zr,

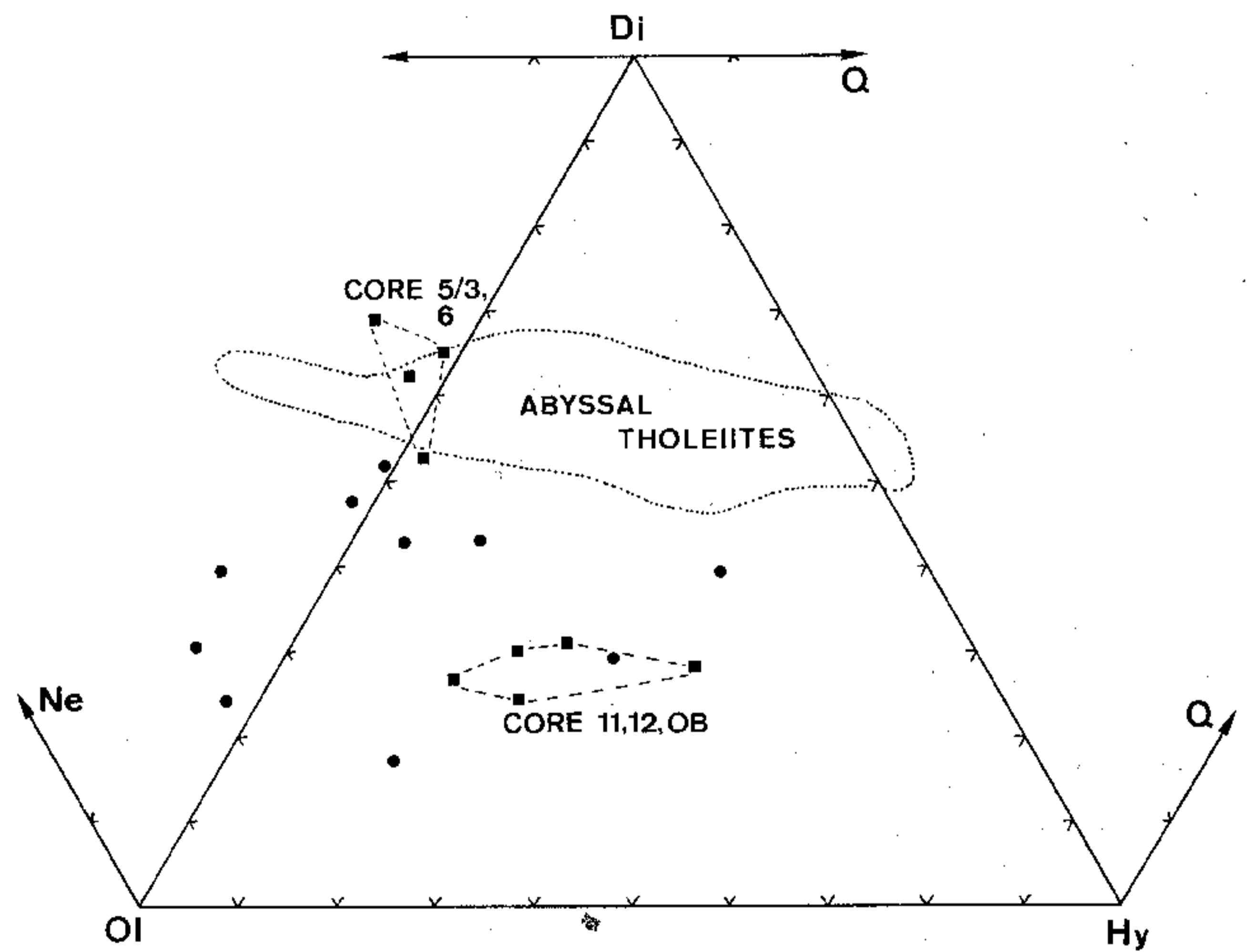


Figure 8. Normative compositions of the Hole 373A basalts plotted in the system Ne-O1-Di-Hy-Q. The encircled area represents the range of abyssal tholeiites from oceanic ridges. Solid circles = low-Ti basalts, solid squares = high-Ti basalts.

TABLE 5
CIPW Norm and Mineralogical Criteria

	5-1, 68-71	5-1, 90-93	5-2, 35-38	5-2, 53.5-57	5-3, 74-75.5	5-3, 129-131	6-1, 96-100							
or	1.89	1.77	1.65	1.89	1.71	2.54	2.25							
ab	20.85	24.79	20.56	23.92	30.65	28.94	38.51							
an	34.07	36.99	36.90	37.30	29.39	23.72	24.67							
ne	2.55	—	—	0.66	1.00	3.67	0.46							
di	5.38	{ 2.77 1.69 .92	11.32	{ 5.84 3.69 1.79	—	13.33	{ 6.89 4.38 2.07	16.88	{ 8.54 4.30 4.05	20.84	{ 10.60 5.71 4.53	17.31	{ 8.70 4.04 4.57	
hy	—	—	12.03	{ 8.09 3.94	8.76	{ 6.02 2.74	—	—	—	—	—	—	—	
ol	17.7	{ 11.08 6.62	6.17	{ 4.02 2.15	14.37	{ 9.57 4.80	15.51	{ 10.20 5.31	11.03	{ 5.42 5.62	11.22	{ 5.98 5.24	9.41	{ 4.19 5.23
mt	1.55	1.42	1.48	1.46	1.91	1.94	1.97							
il	2.13	2.05	2.03	1.96	2.94	3.25	3.32							
ap	0.47	0.40	0.38	0.28	0.62	0.54	0.59							
cc	10.05	—	9.36	—	0.02	—	—							
H ₂ O	3.71	2.49	4.80	2.40	1.88	2.38	0.92							
Total	100.34	99.44	100.36	98.71	98.05	99.04	99.40							
100 Mg Mg+Fe ²⁺	65.32	67.43	69.12	68.42	51.36	55.12	46.15							
Different. Ind.	25.29	26.57	22.22	26.47	33.36	35.15	41.21							
Phenocrysts [size in mm]	plag 1-3	plag	plag 1-2	plag										
Groundmass [plag size in mm]	cpx+plag 0.1-0.3	cpx+plag	cpx+plag 0.1	cpx+plag (albite)	cpx+plag	cpx+plag	cpx+plag							
Calcite [vesicle size in mm]	in veins + vesicles 0.1-1	in vesicles + groundmass 0.1-1												

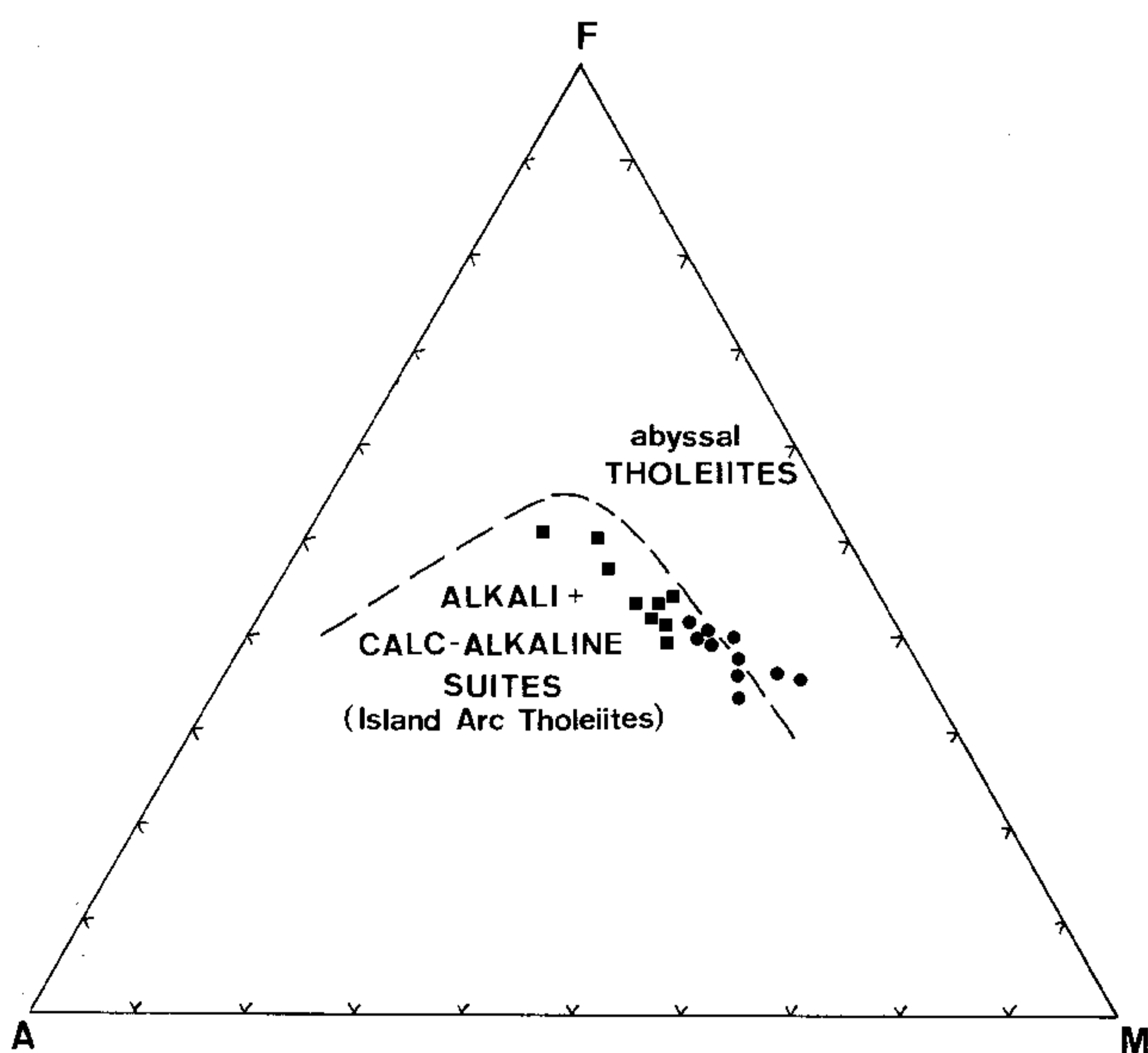


Figure 9. AFM diagram $(Na_2O + K_2O)/FeO_{Total}MgO$ showing the distribution of 20 analyses of the Hole 373A basalts. Solid circles = low-Ti basalts; solid squares = high-Ti basalts.

Y, REE), extensive olivine-pyroxene fractionation, and probably a lack of plagioclase accumulation. But this cannot be further quantified because of the extensive alteration of this group. In conclusion, the trace elements of the Tyrrhenian basalts collected during Leg 42A show low LIL element concentrations similar to abyssal tholeiites. Within this level some elements (K, Rb, Sr, Ba, Pb, U, light REE) have concentrations at the upper limit of the abyssal range, and higher concentrations of these elements in the source rocks. Major element abundances are also similar to those of abyssal tholeiites (Kay et al., 1970; Cann, 1971; Hart, 1971) and to marginal basin basalts (Hart et al., 1972; Ridley et al., 1974; Hawkins, 1976).

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TABLE 5 - Continued

	6, CC	7-1, 139-144	7-2, 136-140	7-3, 111-112.5	7-4, 19-23	7-4, 135-140						
or	1.89	1.99	1.60	1.24	1.42	1.36						
ab	30.55	26.74	22.09	22.88	17.21	19.52						
an	26.78	34.78	42.61	36.79	41.98	37.08						
ne	—	—	—	0.12	3.65	1.02						
di	17.57	{ 9.01 5.37 3.18	12.58	{ 6.49 4.07 2.02	11.89	{ 6.17 4.12 1.59	16.90	{ 8.73 5.55 2.62	8.72	{ 4.52 2.94 1.27	6.26	{ 3.25 2.17 .84
hy												
ol	14.86	{ 8.99 5.88	15.16	{ 9.79 5.37	12.11	{ 8.49 3.61	16.23	{ 10.68 5.55	19.26	{ 13.04 6.22	23.45	{ 16.46 6.99
mt												
il	2.68	1.75	1.60	1.94	1.60	1.73						
ap	.45	0.28	0.24	0.31	0.28	0.26						
cc	—	—	—	—	—	3.41						
H ₂ O	1.97	3.39	2.51	2.41	3.22	4.44						
Total	99.54	99.72	99.40	100.39	98.82	100.09						
$\frac{100 \text{ Mg}}{\text{Mg}+\text{Fe}^2}$	62.94	67.84	72.62	68.82	71.02	73.29						
Different. Ind.	32.44	28.69	23.68	24.24	22.28	21.90						
Phenocrysts [size in mm]			plag 2-3	plag 1-3								
Groundmass [plag size in mm]	cpx+plag	cpx+plag	cpx+plag 0.3-0.5	cpx+plag 0.2-0.4	cpx+plag	cpx+plag						
Calcite [vesicle size in mm]						in vesicles						

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TABLE 5 - Continued

7-5, 10-14	10-1, 123-125	11-1, 114-117	11-1, 130-135	12, CC	OB 2	Ocb
					Outer Barrel	
1.48	1.60	1.60	1.65	1.89	1.89	2.19
19.21	27.42	34.95	36.22	37.23	35.37	37.57
38.73	34.05	25.69	26.08	24.05	26.13	23.45
5.17	8.73	8.39	7.06	7.48	8.43	8.72
5.35	10.01	14.11	7.67	5.21	8.00	7.81
19.90	11.01	7.24	14.02	15.60	11.77	11.61
1.80	1.32	1.86	1.83	1.75	1.83	1.88
2.07	2.11	3.19	2.98	2.96	3.27	3.53
0.33	.33	0.57	0.52	0.52	0.62	0.62
1.82	-	-	-	-	-	-
4.58	3.00	2.27	2.14	2.31	2.12	2.56
100.44	99.57	99.85	100.17	99.01	99.43	99.94
68.79	73.63	61.54	64.61	66.04	62.40	60.96
20.69	29.01	36.54	37.87	39.12	37.26	39.76
	plag 1-2	plag 0.5			none	plag 1-2
cpx+plag	cpx+plag 0.1-0.2	cpx+plag 0.2-0.3	cpx+plag	cpx+plag (albite)	cpx+plag 0.3-0.5	cpx+plag 0.3-0.6
in vesicles	few vesicles	zeolites				
0.1-0.2	0.2-1	0.3-0.4				

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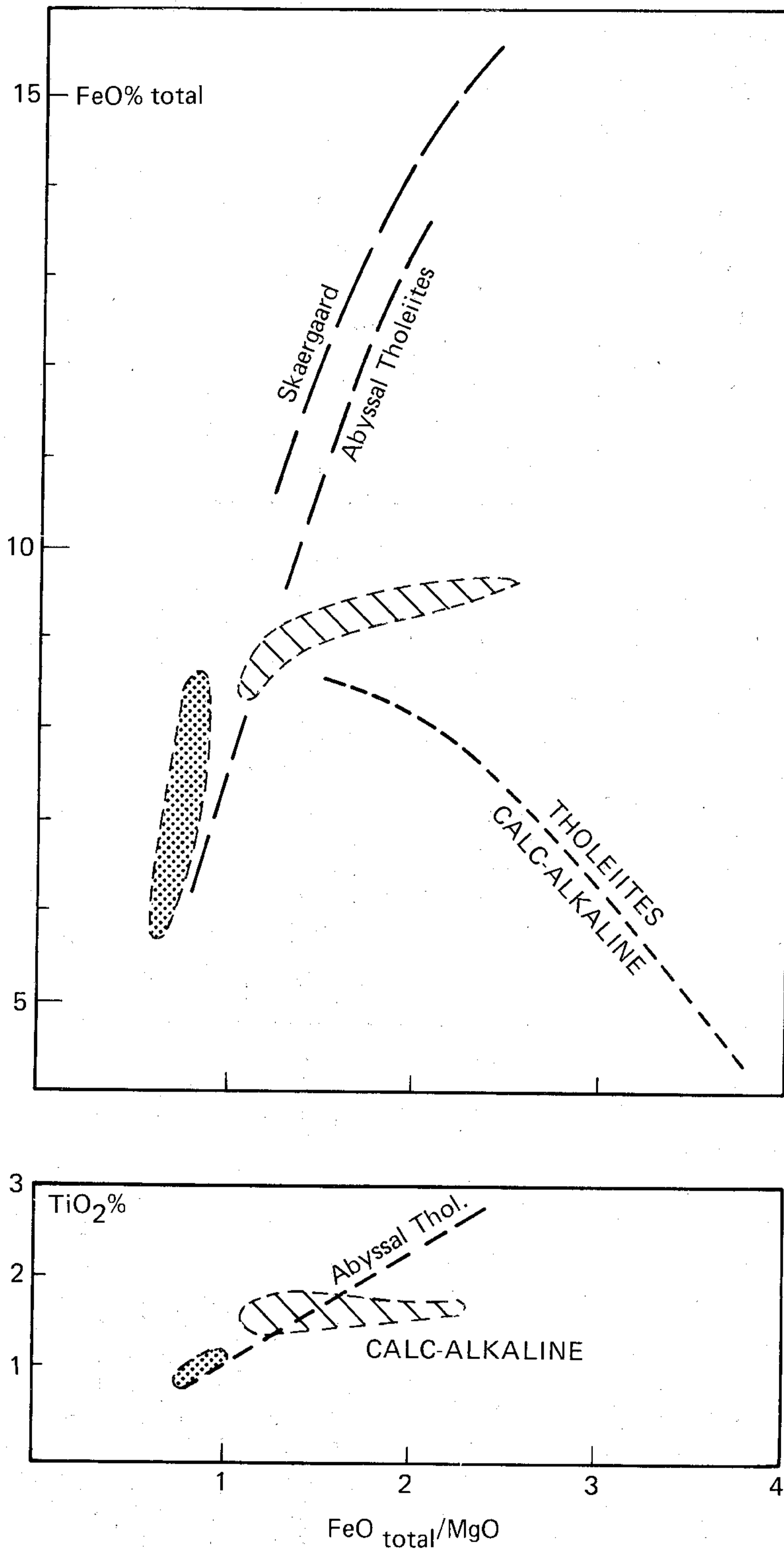


Figure 10. FeO_{Total} and TiO_2 versus FeO_{Total}/MgO of the Hole 373A basalts. The FeO_{Total}/MgO ratio indicates the degree of fractional crystallization (after Miyashiro, 1975). Stippled area = low-Ti basalts; hachured area = high-Ti basalts.

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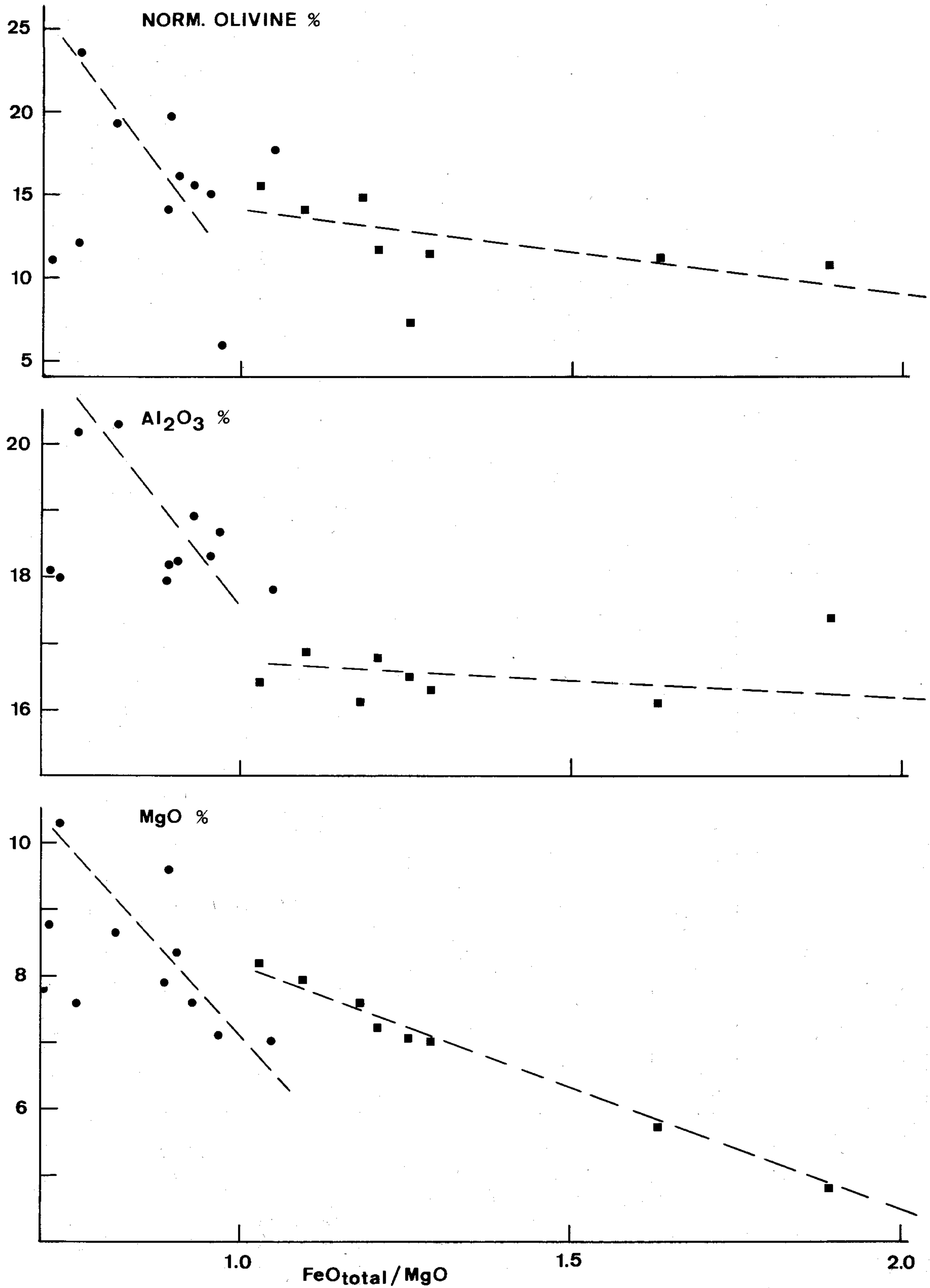


Figure 11. MgO , Al_2O_3 , and normative olivine contents versus FeO_{Total}/MgO . Trend of compositional variation is indicated by straight lines.

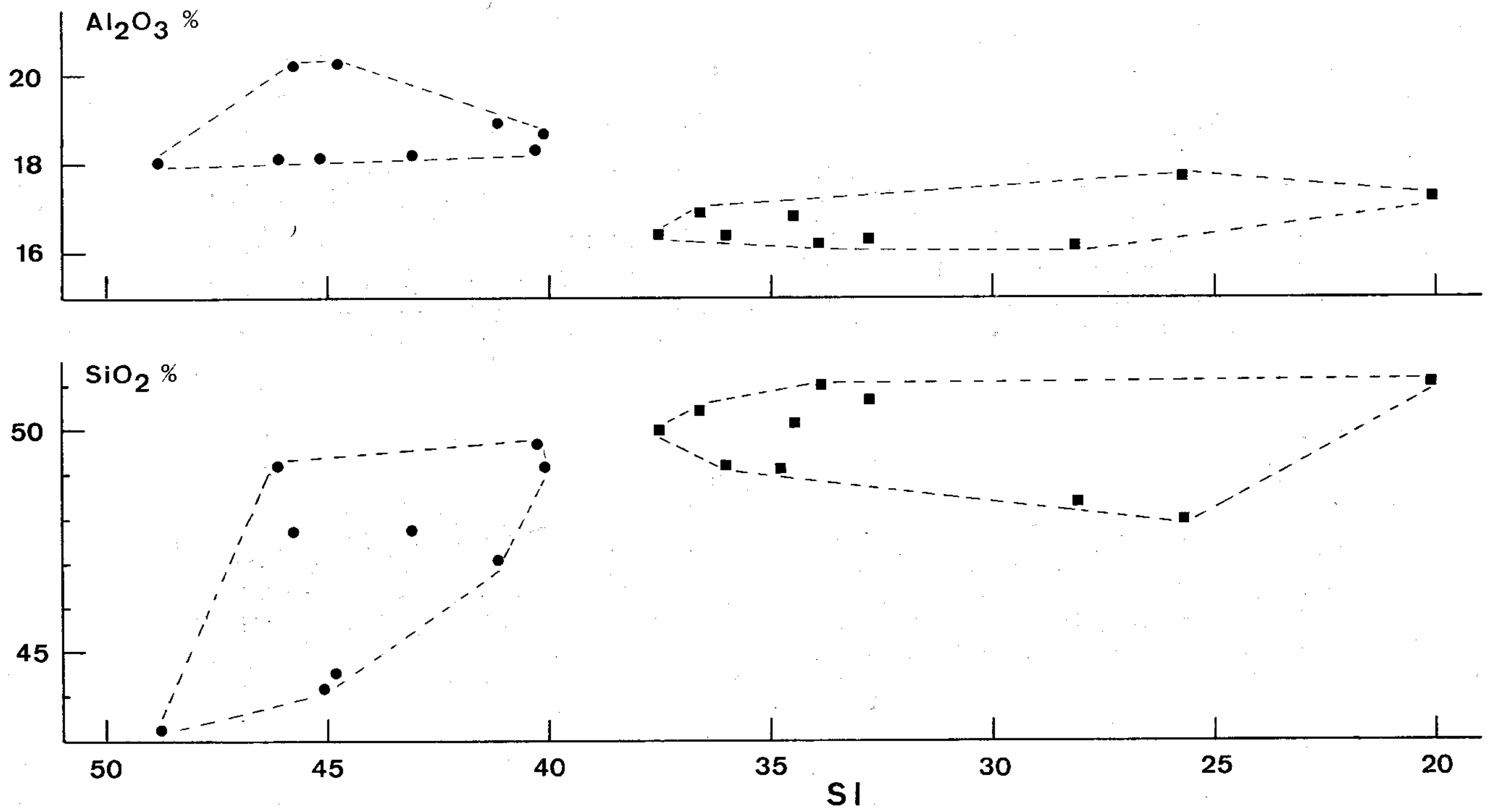


Figure 12. Oxide percents plotted against the solidification index ($SI = \frac{MgO}{MgO + FeO + 0.9Fe_2O_3 + Na_2O + K_2O} \cdot 100$).

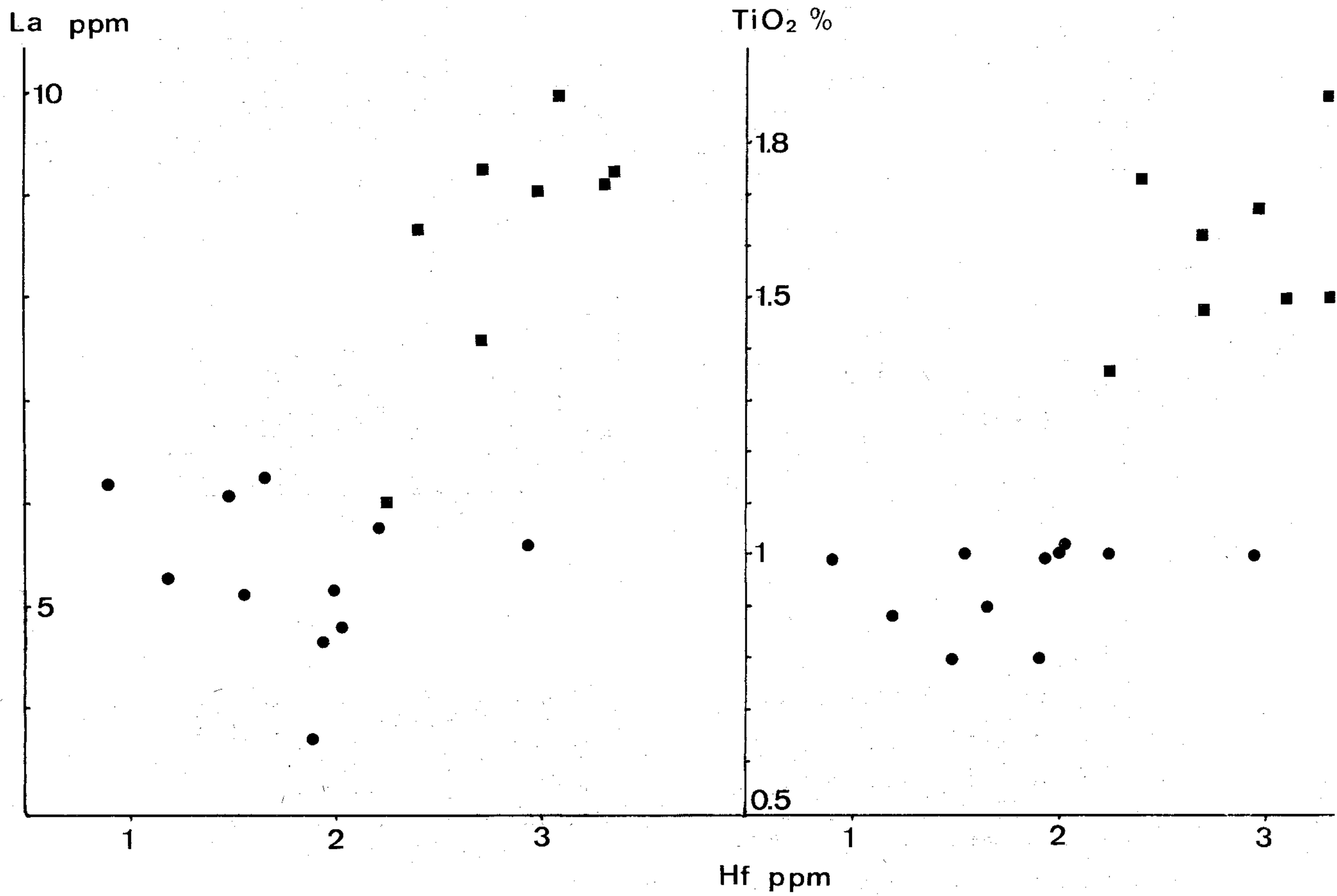


Figure 13. La/Hf and TiO_2/Hf plots for Hole 373A.

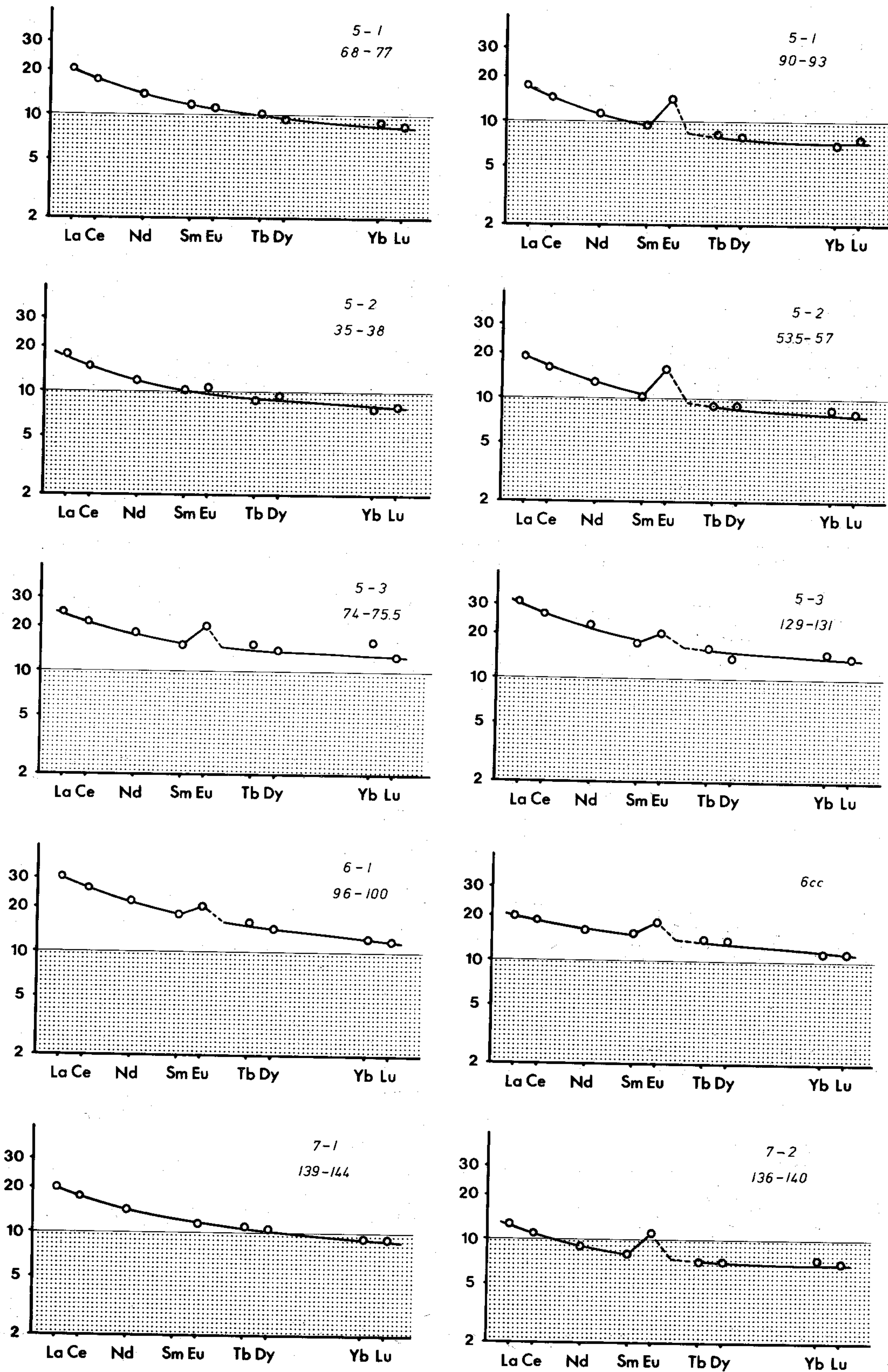


Figure 14. REE pattern of the 20 basaltic samples, Hole 373A.

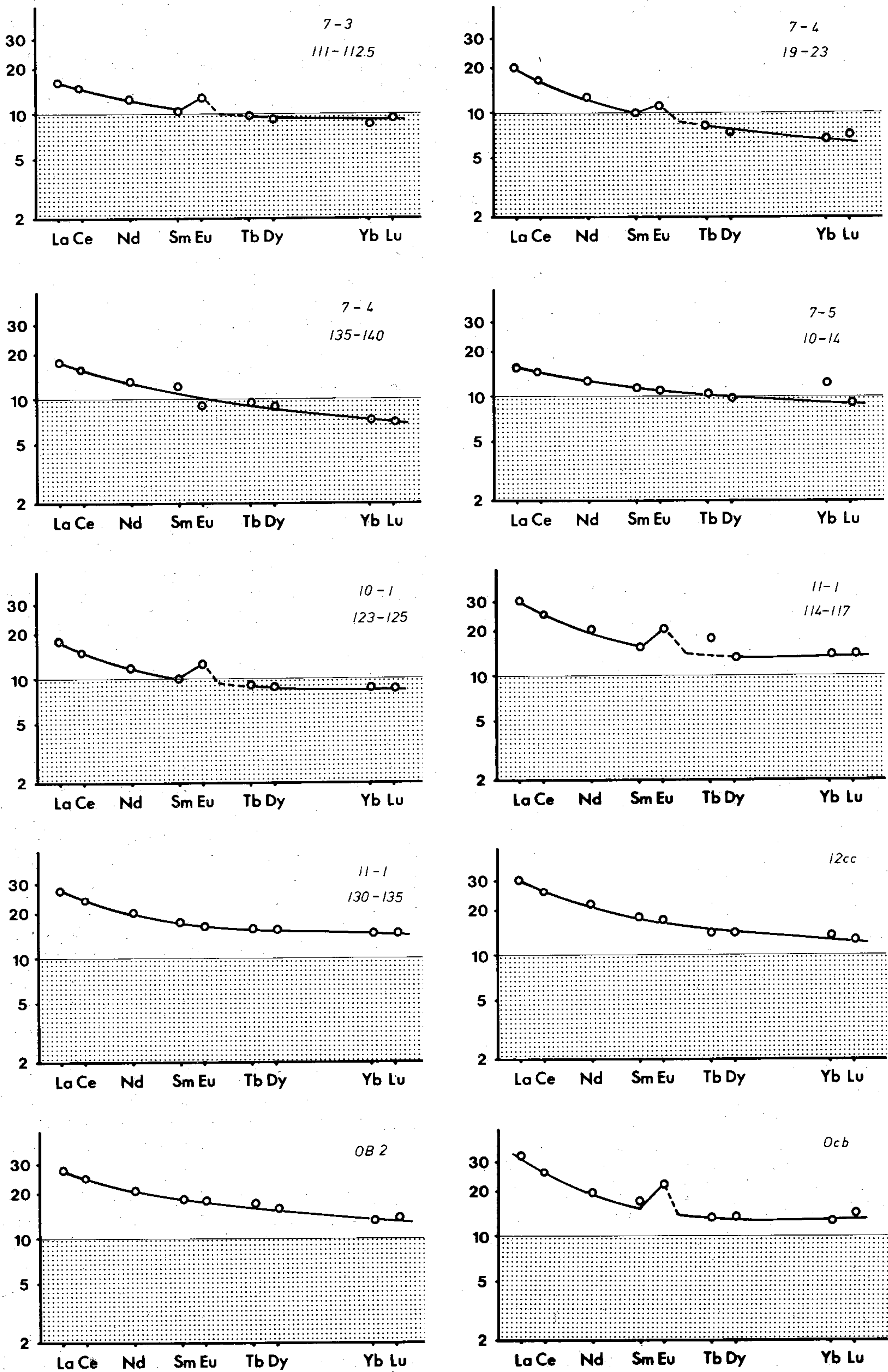


Figure 14. (Continued).